

- 32, 3246 (1967).  
 (10) J. H. Billman and A. C. Diesing, *ibid.*, **22**, 1068 (1957).  
 (11) W. A. White and H. Weingarten, *ibid.*, **32**, 213 (1967).  
 (12) N. J. Leonard and R. R. Sauer, *J. Amer. Chem. Soc.*, **79**, 6210 (1957).  
 (13) J. A. Marshall and W. S. Johnson, *J. Org. Chem.*, **28**, 421 (1963).  
 (14) D. H. R. Barton, A. K. Ganguly, R. H. Hesse, S. M. Loo, and M. M. Pechet, *Chem. Commun.*, 806 (1968).  
 (15) R. Baltzly and A. P. Phillips, *J. Amer. Chem. Soc.*, **68**, 261 (1946).  
 (16) S. S. Tenen, *Psychopharmacologia*, **12**, 1 (1967).  
 (17) F. Sulser, M. H. Bickel, and B. B. Brodie, *J. Pharmacol. Exp. Ther.*, **144**, 321 (1964).

- (18) A. Weissman, *Psychopharmacologia*, **21**, 60 (1971).  
 (19) C. Morpurgo, *Arch. Int. Pharmacodyn.*, **137**, 84 (1962).  
 (20) A. Weissman, S. S. Tenen, and B. K. Koe, *Fed. Proc., Fed. Amer. Soc. Exp. Biol.*, **31**, 529 (1972).  
 (21) C. F. Huebner, U. S. Patent 3,201,470 (1965).  
 (22) J. W. Huffman, *J. Org. Chem.*, **24**, 1759 (1959).  
 (23) R. Ghosh and R. Robinson, *J. Chem. Soc.*, 508 (1944).  
 (24) A. Bhati and N. Kale, *Angew. Chem.*, **79**, 1100 (1967).  
 (25) P. A. S. Smith and W. L. Berry, *J. Org. Chem.*, **26**, 27 (1961).  
 (26) E. Hardegger, E. Redlich, and A. Gal, *Helv. Chim. Acta*, **27**, 628 (1944).  
 (27) L. H. Goodson, I. L. Honigberg, J. J. Lehman, and W. H. Burton, *J. Org. Chem.*, **25**, 1920 (1960).

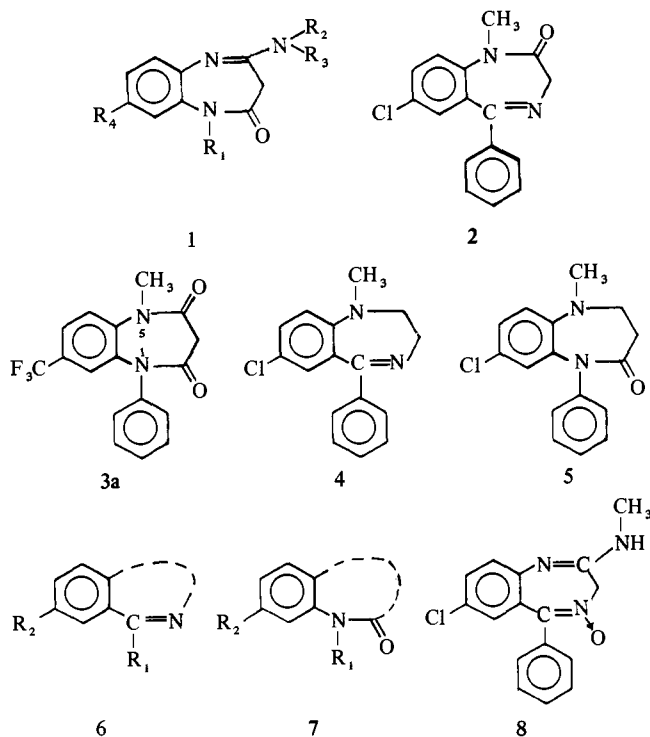
## Benzodiazepines with Psychotropic Activity. 7.<sup>1</sup> Synthesis and Biological Action of 4-Amino-1,5-benzodiazepines

Adolf Bauer,\* Peter Danneberg, Karl-Heinz Weber, and Klaus Minck

Wissenschaftliche Abteilung, C. H. Boehringer Sohn, D-6507 Ingelheim, Federal Republic of Germany. Received December 11, 1972

This paper gives a description of the syntheses of substituted 4-amino-1,5-benzodiazepines **1**. In addition, the possibility is discussed that due to corresponding structural features the 1,5- and 1,4-benzodiazepines (see **6**, **7**) exhibit similar effects on the central nervous system (CNS). Pharmacological data are given for **1**. The biological properties of some particularly active compounds (e.g., **1c** and **1n**) are dealt with in detail.

Diazepam **2**<sup>2</sup> and **3a**<sup>†,3,4</sup> on the one hand and medazepam **4**<sup>2</sup> and benzodiazepines of type **5**<sup>5</sup> on the other hand show a partly analogous action in the animal experiment in respect to their effect on the CNS. Apparently the two structural types **6** and **7** have a similar profile of pharmacological action. Compounds of structure **1** were of interest in this



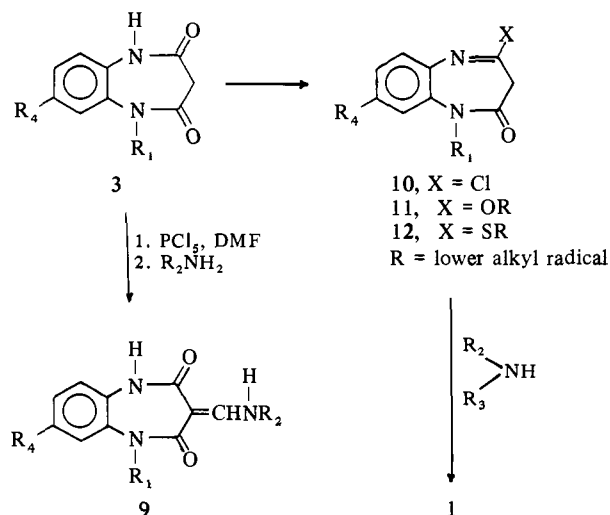
context; they exhibit a structural relationship to chlordiazepoxide<sup>2</sup> **8** and can be expected to have the ability to form water-soluble salts, which appears favorable for pharmacological reasons.

**Chemistry.** 1,5-Benzodiazepine-2,4 diones **3**<sup>4</sup> could be

<sup>†</sup>Compound **3a** has been designated ORF 8063 in earlier publications. In ref 4 it appeared as **1n**.

converted to 3-aminomethylidene-1,5-benzodiazepines **9**<sup>6</sup> by means of phosphorus pentachloride and alkylamines in DMF. A high yield of **9** was obtained only if the alkylamine was added to the reaction mixture of phosphorus pentachloride, DMF, and **3** after several hours. If the alkylamine was added after only a few minutes, the amidine **1** in a mixture of **3** and **9** was obtained. A high yield of **1** was obtained in dioxane and also to some degree in other inert solvent. The imino chloride **10**<sup>7-9</sup> probably occurs as an intermediate but was not isolated. The conversion of **3** to **1** could also be directed through the imino ether **11** or the imino sulfide **12**, which could relatively smoothly be converted to **1** by means of the amine as was expected.<sup>10,11</sup> In order to trace structure-activity relationships we have synthesized a series of these compounds<sup>12</sup> as shown in Scheme I

Scheme I



I. A selection is given together with pharmacological data for guidance (Table I).

Compounds **3**,<sup>4</sup> **5**,<sup>5</sup> **9**,<sup>6</sup> **11**, and **12** have previously been described; we synthesized **3** with  $R_4 = \text{NO}_2$  by way of oxi-

Table I. Compounds of Structure 1, Their Relevant Chemical Data, and Some Pharmacological Properties in Comparison with Chlordiazepoxide

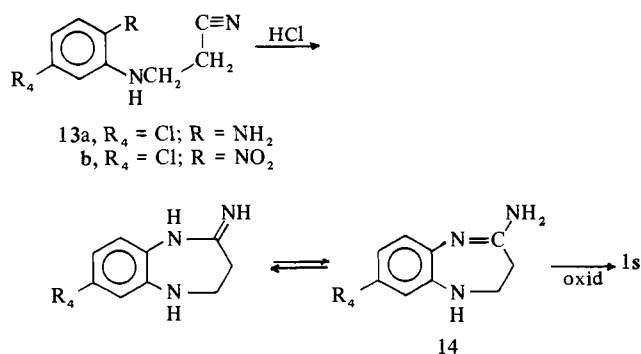
	R <sub>1</sub>	R <sub>2</sub>	R <sub>3</sub>	R <sub>4</sub>	Method <sup>k</sup>	Recrystn solvents	Yield, %	Mp, °C dec	Formula	Mol wt	Analyses	Pharmacology <sup>a</sup>			
												Ataxia, <sup>b</sup> ED <sub>50</sub> , mg/kg	Lying on side, <sup>c</sup> ED <sub>50</sub> , mg/kg	Electroshock, <sup>d</sup> ED <sub>50</sub> , mg/kg	Lethality, <sup>e</sup> LD <sub>50</sub> , mg/kg
<b>Ia</b>	C <sub>6</sub> H <sub>5</sub>	H	H	Cl	A	MeOH- <i>i</i> -Pr <sub>2</sub> O	51	242-243	C <sub>15</sub> H <sub>12</sub> ClN <sub>3</sub> O	285.6	C, H, Cl; N <sup>f</sup>	53	880	32	990
<b>Ib</b>	C <sub>6</sub> H <sub>5</sub>	H	H	CF <sub>3</sub>	A	CH <sub>2</sub> Cl <sub>2</sub> - <i>i</i> -Pr <sub>2</sub> O	54	227-228	C <sub>16</sub> H <sub>12</sub> F <sub>3</sub> N <sub>3</sub> O	319.3	C, H, N, F	126	623	27	2066
<b>Ic<sup>l</sup></b>	C <sub>6</sub> H <sub>5</sub>	H	H	NO <sub>2</sub>	A	DMF	68	238-240	C <sub>16</sub> H <sub>16</sub> N <sub>4</sub> O <sub>6</sub> S	392.4	C, H, N	11	2600	29	3200
<b>Id</b>	C <sub>6</sub> H <sub>5</sub>	H	H	Br	A	CH <sub>2</sub> Cl <sub>2</sub> -Et <sub>2</sub> O	49	248-249 276-278 <sup>l</sup>	C <sub>15</sub> H <sub>12</sub> BrN <sub>3</sub> O	330.2	C, H, N; Br <sup>g</sup>	17	580	37	1300
<b>Ie</b>	C <sub>6</sub> H <sub>5</sub>	H	H	F	A	CH <sub>2</sub> Cl <sub>2</sub> -Et <sub>2</sub> O	43	222-224	C <sub>15</sub> H <sub>12</sub> FN <sub>3</sub> O	269.3	H, N, F; C <sup>h</sup>	105	>807	160	>807
<b>If</b>	<i>o</i> -F-C <sub>6</sub> H <sub>4</sub>	H	H	Cl	A	EtOAc	35	258-259	C <sub>15</sub> H <sub>11</sub> ClFN <sub>3</sub> O	303.8	C, H, N	76	>910	165	>910
<b>Ig</b>	<i>o</i> -Cl-C <sub>6</sub> H <sub>4</sub>	H	H	Cl	A	CH <sub>2</sub> Cl <sub>2</sub> -Et <sub>2</sub> O	38	260-262	C <sub>15</sub> H <sub>11</sub> Cl <sub>2</sub> N <sub>3</sub> O	320.2	C, H, N, Cl	270	1650	100	>2880
<b>Ih</b>	C <sub>6</sub> H <sub>5</sub>	H	<i>i</i> -C <sub>3</sub> H <sub>7</sub>	Cl <sub>3</sub>	A	<i>i</i> -Pr <sub>2</sub> O	42	215-217	C <sub>19</sub> H <sub>18</sub> F <sub>3</sub> N <sub>3</sub> O	361.4	C, H, N	210	1900	88	>3250
<b>Ii</b>	C <sub>6</sub> H <sub>5</sub>	H	<i>t</i> -C <sub>4</sub> H <sub>9</sub>	Cl	A	MeOH	88	275	C <sub>19</sub> H <sub>20</sub> ClN <sub>3</sub> O	341.9	C, H, N, Cl	1750	>3080	>342	>3080
<b>Ij</b>	C <sub>6</sub> H <sub>5</sub>	H	CH <sub>3</sub>	NO <sub>2</sub>	C	CH <sub>2</sub> Cl <sub>2</sub> - <i>i</i> -Pr <sub>2</sub> O	75	217-219	C <sub>16</sub> H <sub>14</sub> N <sub>4</sub> O <sub>3</sub>	310.3	C, N; H <sup>i</sup>	104	>2790	270	>2790
<b>Ik</b>	C <sub>6</sub> H <sub>5</sub>	H	CH <sub>3</sub>	CF <sub>3</sub>	B	CH <sub>2</sub> Cl <sub>2</sub> - <i>i</i> -Pr <sub>2</sub> O	92	201-203	C <sub>17</sub> H <sub>14</sub> F <sub>3</sub> N <sub>3</sub> O	333.3	C, H, N, F	160	660	47	>2700
<b>Il</b>	C <sub>6</sub> H <sub>5</sub>	H	CH <sub>2</sub> CH=CH <sub>2</sub>	Cl	A	THF- <i>i</i> -Pr <sub>2</sub> O	48	173-176	C <sub>18</sub> H <sub>16</sub> ClN <sub>3</sub> O	325.8	C, H, N, Cl	185	>975	185	>975
<b>Im</b>	C <sub>6</sub> H <sub>5</sub>	CH <sub>3</sub>	CH <sub>3</sub>	CF <sub>3</sub>	A	<i>i</i> -Pr <sub>2</sub> O	63	154-155	C <sub>18</sub> H <sub>16</sub> F <sub>3</sub> N <sub>3</sub> O	347.4	C, H, N, F	130	600	41	1200
<b>In</b>	C <sub>6</sub> H <sub>5</sub>	CH <sub>3</sub>	CH <sub>3</sub>	NO <sub>2</sub>	C	CH <sub>2</sub> Cl <sub>2</sub> - <i>i</i> -Pr <sub>2</sub> O	80	219-220	C <sub>17</sub> H <sub>16</sub> N <sub>4</sub> O <sub>3</sub>	324.4	C, H, N	21	>2316	219	>2316
<b>Io</b>	C <sub>6</sub> H <sub>5</sub>	H	CH <sub>3</sub>	Cl	C	CH <sub>2</sub> Cl <sub>2</sub> - <i>i</i> -Pr <sub>2</sub> O	68	217-220	C <sub>16</sub> H <sub>14</sub> ClN <sub>3</sub> O	299.8	C, H, N	23	890	41	>1550
<b>Ip</b>	C <sub>6</sub> H <sub>5</sub>	H	C <sub>2</sub> H <sub>5</sub>	CF <sub>3</sub>	A	<i>i</i> -Pr <sub>2</sub> O	43	181-183	C <sub>18</sub> H <sub>16</sub> F <sub>3</sub> N <sub>3</sub> O	347.4	C, H, N	50	590	44	>3123
<b>Iq</b>	C <sub>6</sub> H <sub>5</sub>	H	(CH <sub>2</sub> ) <sub>2</sub> OH	NO <sub>2</sub>	B	MeOH	67	191-193	C <sub>17</sub> H <sub>16</sub> N <sub>4</sub> O <sub>4</sub>	340.4	C, H; N <sup>j</sup>	140	>3060	>340	>3060
<b>Ir</b>	C <sub>6</sub> H <sub>5</sub>	H	(CH <sub>2</sub> ) <sub>3</sub> OC <sub>2</sub> H <sub>5</sub>	Cl	A	Et <sub>2</sub> O	76	169-170	C <sub>20</sub> H <sub>22</sub> ClN <sub>3</sub> O <sub>2</sub>	371.9	C, H, N, Cl	230	>3200	>360	>3200
<b>Is<sup>m</sup></b>	H	H	H	Cl	<i>n</i>			274-277	C <sub>9</sub> H <sub>8</sub> ClN <sub>3</sub> O	209.6	C, H, Cl, N				
<b>It</b>	C <sub>6</sub> H <sub>5</sub>	H	COCH <sub>3</sub>	Cl	<i>n</i>			226-227	C <sub>17</sub> H <sub>14</sub> ClN <sub>3</sub> O <sub>2</sub>	327.8	C, H, N, Cl	129	>2630	36	>2630
<b>Iu</b>	C <sub>6</sub> H <sub>5</sub>	H	COCH <sub>3</sub>	CF <sub>3</sub>	<i>n</i>			185-187	C <sub>18</sub> H <sub>14</sub> F <sub>3</sub> N <sub>3</sub> O <sub>2</sub>	361.3	C, H, N, F	25	400	58	>1083
<b>Iv</b>	C <sub>6</sub> H <sub>5</sub>	H	COOC <sub>2</sub> H <sub>5</sub>	Cl	<i>n</i>			187-189	C <sub>18</sub> H <sub>16</sub> ClN <sub>3</sub> O <sub>3</sub>	357.8	C, H, N, Cl	290	>3220	68	>3220
<b>8</b>	Chlordiazepoxide											26	540	61	2000

<sup>a</sup>Albino mice (NMRI) of 20-25-g weight were used and also albino rats (FW 49) of 140-200-g weight. The test substances were suspended in olive oil and administered intragastrically by way of an esophageal cannula. The dose in mg/kg being effective in 50% of the animals is stated (ED<sub>50</sub>, LD<sub>50</sub>). These values were obtained graphically from dose-action curves. In all experiments ten animals were used for each dose and the untreated groups. No statistical analysis of the data was employed. <sup>b</sup>Occurrence of ataxia during 8 hr following the application of the drug. <sup>c</sup>Animals lying on their sides, being unable to stand on their feet. The spinal righting reflex still maintains. <sup>d</sup>Electroshock applied by eye electrodes (90 V, 30 mA, 50 Hz sinusoidal, 0.5 sec), prevention of maximum extension seizure: J. E. Toman, E. A. Swinyard, L. S. Goodmann, M. Merkin, and M. Morata, *J. Neurophysiol.*, **9**, 231 (1946). <sup>e</sup>Determination of the lethality within an observation period of 24 hr. <sup>f</sup>N: calcd, 14.74; found, 14.23. <sup>g</sup>Br: calcd, 24.18; found, 23.63. <sup>h</sup>C: calcd, 66.90; found, 65.67. <sup>i</sup>H: calcd, 4.55; found, 5.03. <sup>j</sup>N: calcd, 16.47; found, 16.02. <sup>k</sup>From benzodiazepines 3: A, *via* imino chloride 10; B, *via* imino-ether 11; C, *via* imino sulfide 12. <sup>l</sup>Methanesulfonate. <sup>m</sup>Is was of chemical interest only and not tested pharmacologically. <sup>n</sup>See Experimental Section.

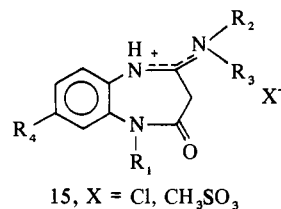
dation<sup>13</sup> from the corresponding substituted tetrahydrobenzodiazepinone of type **5**. In connection with their biochemical properties, these compounds will be dealt with more thoroughly. The iminoether **11** was obtained by conversion of **3** with boron trifluoride etherate and the imino sulfide **12** similarly from **3** by alkylation of the 4-thione prepared by selective thionation of the 4-carbonyl group.<sup>14</sup> Otherwise, substituted compounds **1** were obtained also by reaction of *o*-phenylenediamines with  $\beta$ -chlorocarbonyl- $\alpha$ -chloroanilines.<sup>15</sup>

Compounds **1** with R<sub>1</sub> = H cannot be produced through the same procedure as a further enolizable group is present. Therefore, we synthesized one of these compounds by ring closure of the aminonitrile **13a** to **14** which could be oxidized (**1s**) in a similar way as 1,4-benzodiazepines.<sup>16-18</sup> We obtained **13a** by addition of acrylonitrile<sup>19</sup> to 2-amino-4-chloronitrobenzole in pyridine, in the presence of potassium carbazolate<sup>20</sup> and subsequent reduction of the nitro group of **13b** (see Scheme II).

Scheme II



The amidines **1** are quite thermostable and, with lower alkyl groups R<sub>2</sub> and R<sub>3</sub>, form water-soluble hydrochlorides and methanesulfonates. The structure of the salts apparently coincides with structure **15**.<sup>21</sup> Compounds **1t-v** were ob-



tained from **1a** and **1b** through acylation with either acetic anhydride or ethyl chloroformate. The nmr spectra showed the expected signals in the aliphatic and aromatic regions; e.g., **1k** (in DMSO-*d*<sub>6</sub>)  $\delta$  2.8 (NCH<sub>3</sub>), 3.2 (CH<sub>2</sub>), 6.8-7.5 ppm (arom H). In solutions of **15** the C-3 protons were interchangeable with the deuterium in D<sub>2</sub>O.

**Pharmacology.** Some of the compounds in series **1** inhibit the maximum extensor seizures in mice (Table I). In most compounds this anticonvulsive action is in an order of magnitude similar to that of chlordiazepoxide. The parent substances and the compounds with short side chains are particularly effective. Compared with 1,4-benzodiazepines (e.g., **8**) an interference with motor coordination only occurs at relatively large doses. The high LD<sub>50</sub> indicates a relatively low toxicity, which applies particularly to the 8-nitro-substituted products (e.g., **1c** and **1n**). Table II shows the activity of some selected benzodiazepines of the type **1** compared with chlordiazepoxide **8** in a series of special tests. In mice some of these compounds exert an anxiolytic action and, with considerably higher doses, inhibit exploratory behavior<sup>22</sup> and the locomotion in an open field,<sup>23</sup> they also demonstrate a marked antagonistic effect against pentylenetetrazole,<sup>24</sup> strychnine, and tremorine.<sup>25</sup> However, the Straub effect due to morphine is not inhibited. Furthermore, some of the compounds **1** cause a taming effect in the mink<sup>26</sup> and possess an anxiolytic action in rats exposed to a conflict in a discriminative passive avoidance situation,<sup>27</sup> both tests being rather specific for minor tranquilizers. The lack of effect in the active avoidance test<sup>28</sup> indicates the absence of major tranquilizer properties.

Table II. Pharmacological Profile of Some Compounds of Type 1

Species	Test <sup>a</sup>	1b	1c	1j	1n	1u	1v	8
Mouse	Inhibition <sup>b</sup> of motor coordination (ED <sub>50</sub> )	163	30	>103	16	26	70	15
Mouse	Anxiolysis <sup>c</sup> (DE <sub>50</sub> )	3	17	>310	>324	160	39	7
Mouse	Inhibition <sup>d</sup> of exploration (DE <sub>50</sub> )	200	180	310	135	100	>356	130
Mouse	Inhibition <sup>e</sup> of locomotion (DE <sub>50</sub> )	>1000	750	210	170	>360	290	180
Mink	Taming <sup>f</sup> effect (ED <sub>50</sub> )	80	20	>100	80	50	50	10
Rat	Inhibition of active avoidance <sup>g</sup> (DE <sub>50</sub> )		>135		>135			>40
Rat	Inhibition of passive avoidance <sup>h</sup> (DT <sub>10</sub> )	32	26	>135	96	>135	>135	10.5
Rat	Prevention of max electric shock <sup>i</sup> (ED <sub>50</sub> )	16	140	>103	17	92	>119	37
Mouse	Pentylenetetrazole antagonism <sup>j</sup> (ED <sub>50</sub> )	7	8	>90	6	24	30	3.8
Mouse	Strychnine antagonism <sup>k</sup> (ED <sub>50</sub> )	>100	100		19	33		32
Mouse	Tremorine antagonism <sup>l</sup> (ED <sub>50</sub> )	22	1	>80	1	11	>20	2
Mouse	Morphine antagonism <sup>m</sup> (ED <sub>50</sub> )	>90	>100	>90	>100	>90	>90	25

<sup>a</sup>Albino mice (NMRI) of 20-25-g body weight, albino rats (FW 49) of 140-200-g body weight, and minks from a fur-animal farm were used. The test substances usually were suspended in olive oil; in *f* a 1% solution of hydroxyethylmethylcellulose was used. In all cases intragastric administration was carried out using an esophageal cannula. The effective dose is given in mg/kg. No statistical analysis of the data was employed. <sup>b</sup>Dose at which 50% of the animals slide down on an inclined plane; O. Nieschulz and K. Popendicker, *Arzneim.-Forsch.*, **5**, 458 (1955). <sup>c</sup>Dose at which the locomotion in an open field<sup>23</sup> is increased to 50% indicating an anxiolytic effect. <sup>d</sup>Dose causing a 50% decrease of exploration in the Planché à Trous situation.<sup>22</sup> <sup>e</sup>Dose causing a 50% attenuation of locomotion in the open-field test.<sup>23</sup> <sup>f</sup>Dose eliciting an inhibition of aggressiveness or an occurrence of a taming effect in 50% of the animals.<sup>26</sup> <sup>g</sup>Dose at which the conditioned bar-pressing response necessary to avoid an electroshock punishment is reduced to 50%.<sup>28</sup> <sup>h</sup>Dose at which the animals being in a conflict situation made ten lever presses in order to receive food even though a simultaneous signal indicates the association of the reward with an electric shock punishment.<sup>27</sup> <sup>i</sup>Electroshock applied by eye electrodes (90 V, 30 mA, 50 Hz sinusoidal, 1.0 sec); dose at which the maximum extensor seizure is prevented in 50% of the animals; E. A. Swinyard, W. C. Brown, and L. S. Goodman, *J. Pharmacol. Exp. Ther.*, **106**, 319 (1952). <sup>j</sup>Dose at which the lethal effect of 125 mg/kg of pentylenetetrazole administered intraperitoneally 1 hr after the test substance is prevented in 50% of the animals.<sup>24</sup> <sup>k</sup>Dose at which the lethal effect of 1.9 mg/kg of strychnine sulfate injected intraperitoneally 1 hr after dosing with the test substance is prevented in 50% of the animals. <sup>l</sup>Dose causing a 50% inhibition of the tremor due to subcutaneous injection of 40 mg/kg of tremorine,<sup>25</sup> the degree of tremor being ascertained subjectively and given arbitrary scores. <sup>m</sup>Dose preventing in 50% of the animals the Straub phenomena due to 30 mg/kg of morphine sulfate injected intraperitoneally.

## Discussion

The compounds **1** obviously have similar pharmacological properties as the chemically related **8**. Thus, the assumed pharmacologic analogy has been confirmed. Within the amidine series **1** and also in comparison to **8** differences in their profiles of action have been found. Most effective were the parent substances ( $R_2 = R_3 = H$ ) and those derivatives with short alkyl or acyl chains. Compounds **1** with longer C chains showed a markedly lower effect. Contrary to the benzodiazepinediones **3**, substitution with electro-negative substituents in the ortho position of the phenyl group  $R_1$  caused no increase in effect. To a certain extent the compounds with  $R_4 = Cl$  and  $CF_3$  were more effective than those with  $R_4 = NO_2$  but usually also more toxic. The rate of effectiveness to toxicity was most favorable for the 8-nitro parent substances with short alkyl or acyl chains.

## Experimental Section

The melting points are uncorrected and the yields are not optimized. Ir and nmr spectra were consistent with assigned structures.

4-Amino-1-aryl-2*H*-1,3-dihydro-1,5-benzodiazepin-2-one (**1a-r**) from **3** via the Imino Chloride **10**. General Procedure (Method A).  $PCl_5$  (50 g, 0.24 mol) was added to a solution of 0.03 mol of 1*H*-1,5-benzodiazepine-2,4-(3*H*,5*H*)-dione (**3**) in 750 ml of absolute dioxane and stirred during the process. After reaction for 60 min the suspension was ice cooled and stirred with excess amine. The use of liquid ammonia makes cooling unnecessary; a rapid stream of gaseous amine can also be introduced until the suspension gives an alkaline reaction. The mixture was stirred for another 30 min. Then it was evaporated *in vacuo* and the residue stirred with cold aqueous  $NH_3$ . This was shaken with  $CH_2Cl_2$ , washed with  $H_2O$ , dried with  $MgSO_4$ , and again evaporated. The residue was recrystallized from MeOH,  $CH_2Cl_2$ , EtOAc, (*i*-Pr) $_2O$ , or mixtures thereof. In preparations of **1** in which  $R_2 = R_3 = H$  it is practicable to take up the residue in absolute  $Me_2CO$  and to separate the product from unaltered starting material by precipitation with  $Et_2O-HCl$ . The base can then be released with aqueous  $NH_3$  and recrystallized.

**1** from Ether **11** or Thioether **12** (Methods B and C, respectively). Either 0.02 mol of 4-ethoxy-2*H*-1,3-dihydro-1,5-benzodiazepin-2-one (**11**) or 4-methylmercapto-2*H*-1,3-dihydro-1,5-benzodiazepin-2-one (**12**) was suspended or dissolved (according to solubility) in 60 ml of EtOH and 5 ml of DMSO. Excess amine was then added and heating was carried out for 2–3 hr. An autoclave was used with gaseous amines or they were led into the solution boiling under reflux. After cooling and evaporation *in vacuo* the residue was extracted with  $CH_2Cl_2$ , washed with  $H_2O$ , dried over  $MgSO_4$ , and again evaporated. The residue was recrystallized (according to solubility) from EtOH, EtOAc, (*i*-Pr) $_2O$ , or their mixtures.

3-(5-Chloro-2-nitrophenylamino)propionitrile (**13b**). Acrylonitrile (11 g, 0.205 mol) was poured quickly into a mixture of 30 g (0.174 mol) of 5-chloro-2-nitroaniline, 30 ml of pyridine, and 0.6 g of potassium carbazolate while being stirred. After heating at 70° for 30 min (*i*-Pr) $_2O$  was added, and the precipitate was filtered by suction and washed with (*i*-Pr) $_2O$ : yield, 29.5 g (75.4%); mp 152–154°.

3-(5-Chloro-*o*-phenylenediamino)propionitrile (**13a**). **13b** (10 g, 0.0445 mol) was dissolved in MeOH and reduced with Raney nickel- $H_2$  at room temperature and 5 atm of pressure. After completion of hydrogen uptake it was passed over kieselguhr and evaporated *in vacuo* and the residue was recrystallized from a little  $CH_2Cl_2-Et_2O$ : yield, 5.46 g (63%); mp 87–89°.

4-Amino-8-chloro-1*H*-2,3-dihydro-1,5-benzodiazepine (**14**). Saturated ethereal HCl was mixed with a solution of 2 g (0.0102 mol) of **13a** in 10 ml of absolute THF. Dry HCl gas was then passed through for 2 hr and the composition was allowed to stand for 15 hr at room temperature. Absolute  $Et_2O$  was then added carefully and the precipitate filtered by suction and dissolved in  $H_2O$ . This was then made alkaline with 6*N* NaOH and repeatedly shaken with  $CH_2Cl_2$ . The  $CH_2Cl_2$  phase was washed with  $H_2O$ , dried with  $MgSO_4$ , and filtered by suction over kieselguhr. After evaporating *in vacuo* the residue was recrystallized from a little  $CH_2Cl_2-(i-Pr)_2O$ : yield, 1.3 g (65%); mp 143–144°. *Anal.* ( $C_9H_{10}ClN_3$ , mol wt 195.7) C, H, N, Cl.

4-Amino-8-chloro-2*H*-1,3-dihydro-1,5-benzodiazepin-2-one (**1s**).

**14** (5 g, 0.0264 mol) was dissolved in 500 ml of  $Me_2CO$  and in small portions mixed with 25 ml of chromic acid (2.67 g of  $CrO_3$  and 2.3 ml of concentrated  $H_2SO_4$  diluted with  $H_2O$  to 10 ml). The mixture was stirred at room temperature for 5 hr. It was then poured into ice-water, neutralized with 2*N* NaOH, shaken with  $CH_2Cl_2$ , dried with  $MgSO_4$ , and evaporated *in vacuo*. The residue was recrystallized from  $Me_2CO$  and dried over  $P_2O_5$ , yield 1.7 g (31.7%).

4-Acetylamino-1-phenyl-2*H*-1,3-dihydro-1,5-benzodiazepin-2-ones (**1t,u**). Either **1a** or **1b** (0.01 mol) was heated under reflux with 7 ml of  $Ac_2O$  in 150 ml of absolute  $C_6H_6$ . After evaporation *in vacuo* the residue was stirred with water and repeatedly extracted with  $CH_2Cl_2$ . After drying with  $MgSO_4$  the solvent was evaporated and the residue recrystallized from (*i*-Pr) $_2O$ . The yields were 85 and 91%.

4-Ethoxycarbonylamino-8-chloro-1-phenyl-2*H*-1,3-dihydro-1,5-benzodiazepin-2-one (**1v**). **1a** (0.01 mol) was dissolved in 150 ml of absolute  $C_6H_6$  and 10 ml of pyridine and mixed with 0.012 mol of ethyl chloroformate. It was allowed to react under ice cooling for 30 min and then stirred into ice-water. After neutralization with 2*N* HCl, it was extracted with EtOAc. Finally the solvent was dried with  $MgSO_4$  and evaporated *in vacuo*, and the residue was crystallized from EtOAc-(*i*-Pr) $_2O$ , yield 68%. Melting points and analytical data of the compounds **1s-v** are given in Table I.

**Acknowledgment.** We extend our thanks to Dr. K. H. Pook for the processing and discussion of the spectra and Mr. M. Unruh and Mrs. W. Gottschall for their skillful assistance.

## References

- (1) K.-H. Weber and A. Bauer, *Justus Liebig's Ann. Chem.*, **763**, 66 (1972) (paper 6).
- (2) L. H. Sternbach, *Angew. Chem.*, **83**, 70 (1971).
- (3) R. D. Heilman, R. J. Matthews, G. O. Allen, and J. P. Da Vanzo, *Clin. Res.*, **19**, 714 (1971).
- (4) K.-H. Weber, A. Bauer, and K. H. Hauptmann, *Justus Liebig's Ann. Chem.*, **756**, 129 (1972).
- (5) A. Bauer, K.-H. Weber, and M. Unruh, *Arch. Pharm. (Weinheim)*, **305**, 557 (1972).
- (6) A. Bauer, K.-H. Pook, and K.-H. Weber, *Justus Liebig's Ann. Chem.*, **757**, 87 (1972).
- (7) Sandoz AG (to R. G. Griot), German Offen. 1,620,360 (1970).
- (8) J. W. Williams, C. H. Witten, and J. A. Krynsky in "Organic Syntheses," Collect. Vol. III, Wiley, New York, N. Y., 1955, p 818.
- (9) W. Kantlehner and P. Speh, *Chem. Ber.*, **104**, 3714 (1971).
- (10) R. J. Morath and G. W. Stacy in "The Chemistry of the Carbon-Nitrogen Double Bond," S. Patai, Ed., Interscience, New York, N. Y., 1970, pp 224 ff.
- (11) Hoffmann La Roche AG (to G. A. Archer and L. H. Sternbach), German Offen. 1,695,217 (1972).
- (12) C. H. Boehringer Sohn (to A. Bauer, K.-H. Weber, K. Minck, and P. Danneberg), Belgium Patent 774,873, German Offen. 2,053,680 (1972); *Chem. Abstr.*, **76**, 48,527u (1972).
- (13) C. H. Boehringer Sohn (to K.-H. Weber and A. Bauer), Belgium Patent 762,901, German Offen. 2,006,601 (1971).
- (14) K.-H. Weber and A. Bauer, *Justus Liebig's Ann. Chem.*, in press.
- (15) R. Buyle and H. G. Viehe, *Tetrahedron*, **25**, 3453 (1963).
- (16) R. I. Fryer, G. A. Archer, B. Brust, and L. H. Sternbach, *J. Org. Chem.*, **30**, 1308 (1965).
- (17) A. M. Felix, J. V. Earley, R. I. Fryer, and L. H. Sternbach, *J. Heterocycl. Chem.*, **5**, 731 (1968).
- (18) K. H. Wünsch, H. Dettmann, and S. Schönberg, *Chem. Ber.*, **102**, 3891 (1969).
- (19) O. Bayer, *Angew. Chem.*, **61**, 229 (1949).
- (20) Houben-Weyl, "Die Methoden der organischen Chemie," Vol. 11/1, Georg Thieme Verlag, Stuttgart, 1957, p 275.
- (21) H. Möhrle, D. Schittenhelm, and P. Gundlach, *Arch. Pharm. (Weinheim)*, **305**, 108 (1972).
- (22) J.-R. Borssin, P. Simon, and J.-M. Lwoff, *Therapie*, **19**, 571 (1964).
- (23) P. A. J. Janssen, A. H. M. Jagenau, and K. H. L. Schellekens, *Psychopharmacologia*, **1**, 389 (1960).
- (24) M. I. Gluckmann, *Curr. Ther. Res.*, **7**, 721 (1965).
- (25) G. M. Everett, *Nature (London)*, **177**, 1238 (1956).
- (26) A. Bauen and G. J. Possanza, *Arch. Int. Pharmacodyn.*, **186**, 133 (1970).
- (27) J. Geller, *ibid.*, **149**, 243 (1964).
- (28) P. B. Dobrin and R. L. Rhyne, *ibid.*, **178**, 351 (1969).